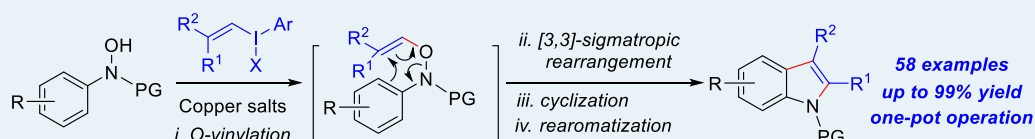


Copper-Catalyzed Tandem O-Vinylation of Arylhydroxylamines/[3,3]-Rearrangement/Cyclization: Synthesis of Highly Substituted Indoles and Benzoindoles

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Supporting Information



ABSTRACT: Herein, we developed a copper-catalyzed O-vinylation of arylhydroxylamine using vinylidonium salts as vinylation reagents to generate a transient O-vinyl-N-arylhydroxylamine that rapidly undergoes a [3,3]-sigmatropic rearrangement and subsequent cyclization/rearomatization to form a substituted indole. A wide range of highly substituted indoles and benzoindoles can be afforded in good yields. This approach is readily scalable, and the scope and application of this process are presented.

KEYWORDS: copper, arylhydroxylamines, vinylidonium salts, rearrangement, indoles

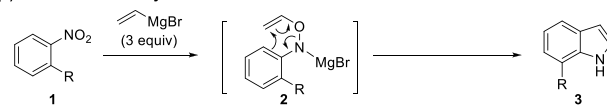
The indole nucleus is one of the most ubiquitous scaffolds because of its wide presence in a plethora of natural products, pharmaceuticals, and agrochemicals, as well as in materials science.¹ In particular, highly substituted indoles have been referred to as “privileged structures” because of their capability of binding to a variety of receptors with high affinity.² In view of the importance and abundance of the indole motif, it is not surprising that significant efforts have been devoted to develop new strategies for the generation of indole units, and numerous methods have been reported^{1g,3} (for example, Fischer,⁴ Madelung,⁵ Hegedus,⁶ Bartoli,⁷ Larock,^{3c,8} and Buchwald⁹ indole synthesis and so on). However, via these methods, harsh conditions (for example, high temperature (>100 °C), low temperature (<−40 °C)), specific starting material availability, and low functional-group tolerance often hamper the versatility and utility of indole synthesis. Thus, the continuous development of alternative approaches that may allow for the straightforward construction of structurally diverse indoles (in particular, 3-substituted indoles) is still a field of increasing interest.

In 1989, Bartoli and co-workers described the reaction of *ortho*-substituted nitroarenes **1** with an excess (3 equiv or more) of vinyl Grignard reagents at low temperature to generate 7-substituted indoles **3** upon aqueous workup conditions.^{7c} The presumably formed intermediate **2** from the addition of the second equivalent of Grignard reagent to the corresponding nitrosoarene, which was generated in situ by the addition of the first equivalent of Grignard reagent to the oxygen of the nitro group followed by the rapid elimination/decomposition of the O-alkenylated intermediate, could undergo a facile [3,3]-sigmatropic rearrangement, followed by an intramolecular nucleophilic addition and rearomatization

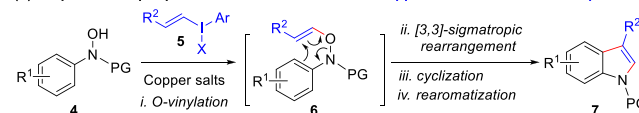
to furnish the final indole products (Scheme 1a). This approach has been proven successful in a number of synthetic

Scheme 1. Proposed Indole Synthesis Inspired by the Bartoli Reaction

(a) The Bartoli indole synthesis: Restricted to 7-substituted indoles.



(b) Proposed one-pot process to substituted indoles: Applicable to all substitution patterns.



applications involving a series of bioactive molecules.^{7e,10} While powerful, this method is inherently limited to substrates which have to possess a substituent *ortho* to the nitro group of the nitroarenes; otherwise, the reaction gives low or no yield of the desired indole product. In addition, 3 equiv of the alkenyl Grignard reagent and harsh conditions of low temperature (−78 to −20 °C) are necessary. The products of the Bartoli indole synthesis are restricted to 7-substituted indoles, and the yields are usually moderate (less than 70%). If such a process could be generalized for non-*ortho*-substituted nitroarene

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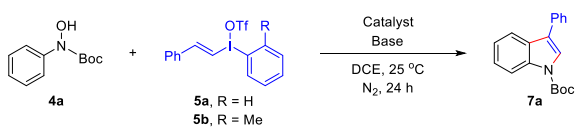


substrates, the scope of the potential applications could be further expanded to a broader range of transformations.

Along this line and inspired by the Bartoli indole synthesis, we surmised that the direct *O*-vinylation of arylhydroxylamines **4** would occur using highly reactive vinyliodonium salts **5** as the vinylation reagent to form *O*-vinyl-*N*-arylhydroxylamines **6**, which would rapidly undergo a similar [3,3]-rearrangement/cyclization/rearomatization to afford indole products **7** (Scheme 1b). This approach utilizes *N*-protected arylhydroxylamines, which are readily prepared from nitroarenes,¹¹ as substrates to undergo a copper-catalyzed cross-coupling reaction¹² with vinyliodonium salts. Vinyliodonium salts are an environmentally benign electrophilic vinylation reagent with low toxicity, high reactivity, and moisture and air stability.¹³ However, in comparison to the diverse utility of analogous diaryliodonium salts¹⁴ in organic synthesis, vinyliodonium salts have been much less explored.

Initially, we chose *N*-Boc-phenylhydroxylamine (**4a**) and (*E*)-phenyl(styryl)iodonium trifluoromethanesulfonate (**5a**) as model substrates to optimize the reaction conditions (selected results are summarized in Table 1 and detailed optimization

Table 1. Optimization of the Reaction Conditions^a



entry	5a/5b	base	catalyst	yield of 7a, % ^b
1	5a	Na ₂ CO ₃		0
2	5a	<i>t</i> BuOK		0
3	5a	Na ₂ CO ₃	Cu(OTf) ₂	0
4	5a	Na ₂ CO ₃	Cu(OAc) ₂	0
5	5a	Na ₂ CO ₃	CuI	trace
6	5a	Na ₂ CO ₃	CuCl	34
7	5a	Na ₂ CO ₃	CuBr	50
8	5a	K ₂ CO ₃	CuBr	29
9	5a	CS ₂ CO ₃	CuBr	34
10	5a	DTBP	CuBr	0
11	5b	Na ₂ CO ₃	CuBr	60
12 ^c	5b	Na ₂ CO ₃	CuBr	64
13 ^d	5b	Na ₂ CO ₃	CuBr	73
14 ^e	5b	Na ₂ CO ₃	CuBr	78

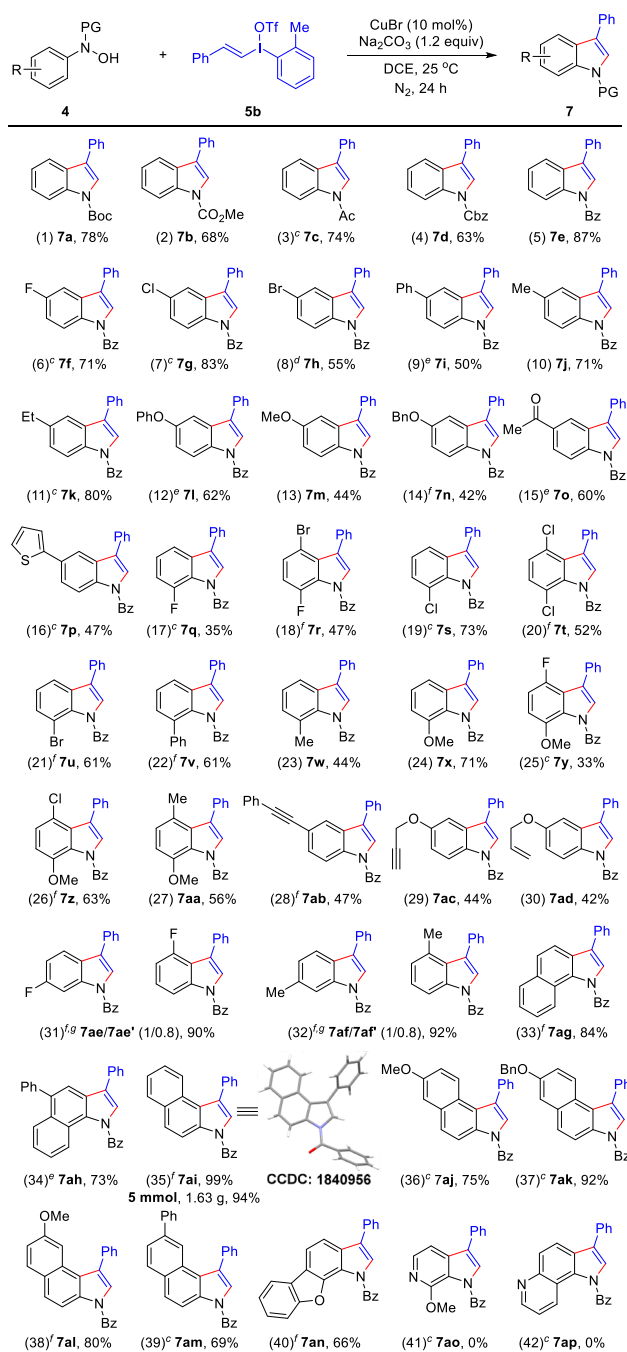
^aUnless otherwise noted, all reactions were carried out under the following conditions: **4a** (0.2 mmol), **5a** or **5b** (1.1 equiv), base (1.5 equiv), catalyst (10 mol %), DCE (1 mL) at 25 °C under N₂ for 24 h. Abbreviations: Boc = *tert*-butyloxycarbonyl; DCE = 1,2-dichloroethane; DTBP = 2,6-di-*tert*-butylpyridine; Tf = trifluoromethanesulfonyl; Ac = acetyl. ^bYields of isolated products. ^c1.5 equiv of **5b** was employed. ^d2.0 equiv of **5b** was employed. ^e1.2 equiv of base was employed.

results are summarized in the Supporting Information). Without the employment of catalyst, the reaction did not occur (Table 1, entries 1 and 2). The screening of copper salt catalysts revealed that CuBr was an efficient catalyst to afford a 50% yield of the desired indole product in the presence of Na₂CO₃ (Table 1, entries 3–7). Various bases (K₂CO₃, CS₂CO₃, DTBP) were also tested, and no higher yields were provided (Table 1, entries 8–10). To our delight, when (*E*)-styryl(*o*-tolyl)iodonium trifluoromethanesulfonate (**5b**) was used as the vinylation reagent, a higher yield was afforded (Table 1, entry 11). Further optimization screening showed

that increasing the loading of iodonium salt **5b** had a positive effect on the reaction (Table 1, entries 12 and 13). The results revealed that 2.0 equiv of **5b**, 1.2 equiv of Na₂CO₃, and 10 mol % CuBr in DCE at 25 °C were optimal conditions (Table 1, entry 14).

With the optimized conditions in hand, we next turned our attention to assessing the scope and limitations of this transformation. We were pleased to find that the copper-catalyzed cascade *O*-vinylation¹⁵/rearrangement/cyclization works across a broad range of arylhydroxylamines, providing access to a diverse array of substituted indole motifs (Table 2). We first explored the scope of the protecting group on the nitrogen atom and found that the benzoyl group is the best option to give a good yield of the desired indole product (Table 2, entries 1–5). The variation of different substituents at the *para* position of the phenyl group was then examined. Both electron-withdrawing groups and electron-donating groups can be well tolerated in this transformation to afford the corresponding indole products in moderate to good yields (Table 2, entries 6–14). It is noteworthy that the acetyl and 2-thiophenyl groups are compatible with this reaction system (Table 2, entries 15 and 16). Meanwhile, various *ortho*-substituted arylhydroxylamines were amenable to the optimized reaction conditions to generate 7-substituted indoles (Table 2, entries 17–27). Notably, different disubstituted aromatic rings, in particular, dihalide-substituted substrates, were also well tolerated in this transformation (Table 2, entries 18, 20, and 25–27). Probably, the steric hindrance of the *ortho* substituents has a negative effect on the efficiency of the reaction and resulted in relatively lower yields in comparison to the *para*-substituted substrates (Table 2, entries 6 vs 17, 7 vs 19, 10 vs 23). To our delight, a series of substrates with redox-sensitive moieties, such as alkynes and olefins, can also be well tolerated (Table 2, entries 28–30). When *meta*-substituted arylhydroxylamines were used as substrates, low regioselectivities but good yields were observed (Table 2, entries 31 and 32). Furthermore, this methodology is applicable to more complex aromatic rings, such as the naphthalene system (Table 2, entries 33–39) and dibenzofuran system (Table 2, entry 40); especially noteworthy are the excellent regioselectivities and high isolated yields of benzoindoles when 2-substituted naphthylhydroxylamines were employed as substrates (Table 2, entries 35–39). The structure of the product was unambiguously confirmed by the single crystal X-ray diffraction study of compound **7ai** (Table 2, entry 35).¹⁶ However, this method is not suitable to pyridine- and quinoline-containing substrates, due to the favored coordination of nitrogen atom to copper preventing the formation of the highly electrophilic vinyl copper complex which was generated in situ between vinyliodonium salts and the copper catalyst (Table 2, entries 41 and 42).

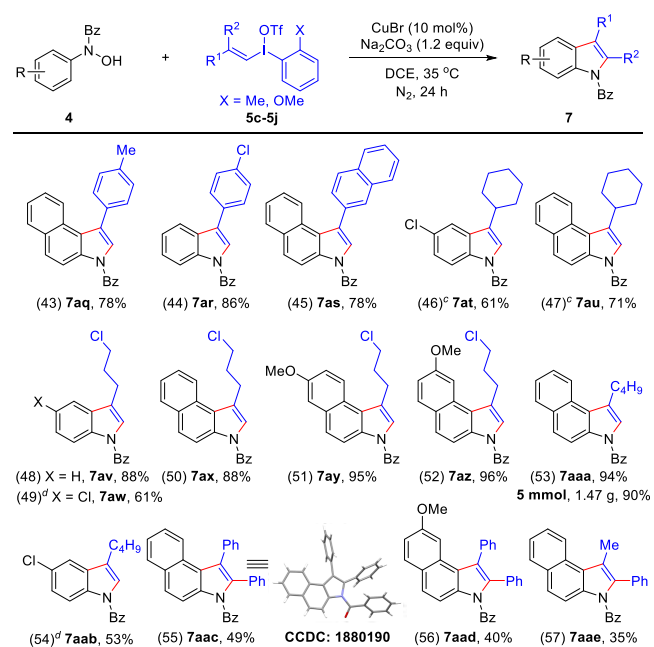
Further investigation with respect to the scope of vinyliodonium salts was conducted (Table 3). As shown in Table 3, these optimized conditions are amenable to a wide range of alkenyliodonium triflates. Both electron-rich and electron-poor styrenes (Table 3, entries 43–45) as well as alkylvinyl groups (Table 3, entries 46–54) are efficiently transferred onto the indole motif with good to excellent yields. Notably, 2,3-disubstituted benzoindoles **7aac**–**7aae** can also be synthesized in moderate yields employing (2,2-diphenylvinyl)(*o*-tolyl)-λ³-iodanyl trifluoromethanesulfonate (**5i**) and (2-methoxyphenyl)(2-phenylprop-1-en-1-yl)-λ³-iodanyl trifluoromethanesulfonate (**5j**) as electrophiles under standard

Table 2. Substrate Scope of Arylhydroxylamines^{a,b}

^aReaction conditions unless specified otherwise: **4** (0.2 mmol), **5b** (0.4 mmol), CuBr (10 mol %), Na₂CO₃ (0.24 mmol), DCE (1 mL) at 25 °C under N₂ for 24 h. Abbreviations: Cbz = benzoxycarbonyl; Bz = benzoyl. ^bYields of isolated products. ^cAt 50 °C. ^dAt 70 °C. ^eAt 60 °C. ^fAt 35 °C. ^gThe regioselectivity was determined by ¹H NMR analysis of the crude reaction mixture; combined yield of the pure regioisomers.

conditions (Table 3, entries 55–57). The structure of the product **7aac** was also confirmed by the single-crystal structure, which was analyzed by X-ray diffraction.¹⁶

To demonstrate the synthetic utility of this one-pot (benzo)indole process on a multigram scale, we chose naphthylhydroxylamine **4ai** and vinyliodonium salts **5b,h** as substrates. We gratefully found that benzoindole **7ai,aaa** were

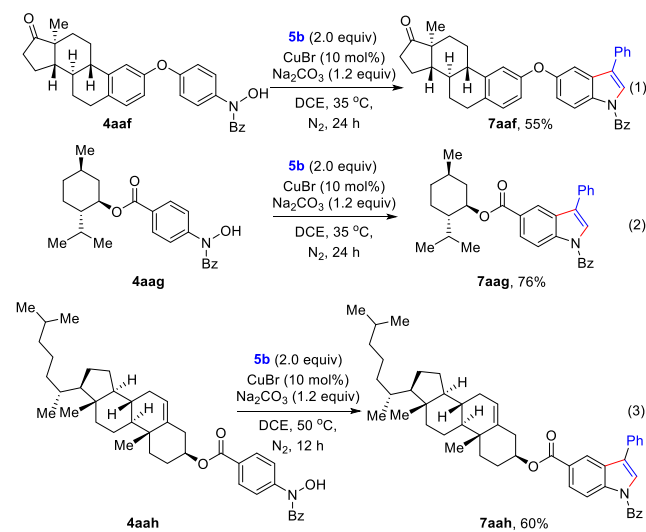
Table 3. Substrate Scope of Vinylidonium Salts^{a,b}

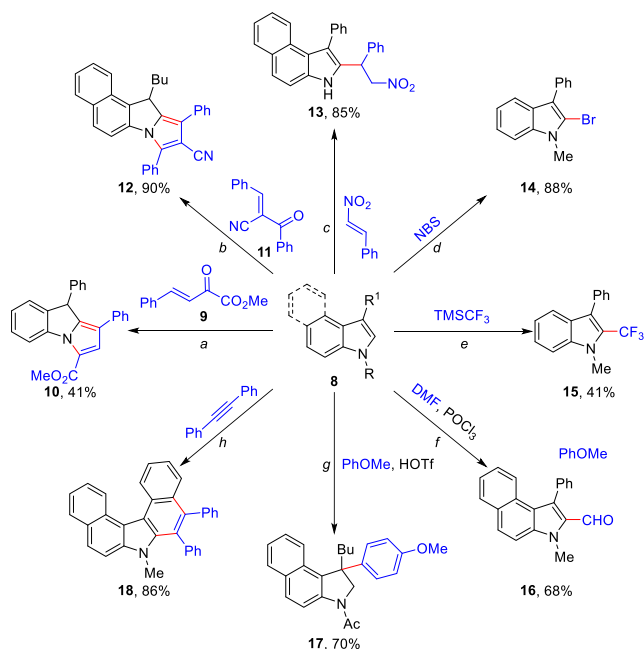
^aReaction conditions unless specified otherwise: **4** (0.2 mmol), **5** (0.4 mmol), CuBr (10 mol %), Na₂CO₃ (0.24 mmol), DCE (1 mL) at 35 °C under N₂ for 24 h. ^bYields of isolated products. ^cAt 60 °C. ^dAt 50 °C.

afforded in 94% and 90% isolated yields with excellent regioselectivities, respectively (Table 2, entry 35 and Table 3, entry 53). This methodology can also be applied to the late-stage functionalization of pharmaceutically relevant and structurally complex intermediates, such as the estradiol derivative **7aaf** (Scheme 2, eq 1), the terpenoid derivative **7aag** (Scheme 2, eq 2), and the cholesterol derivative **7aah** (Scheme 2, eq 3).

To further expand the synthetic applications of our indole process, we wished to demonstrate that the products of this reaction can be selectively manipulated to more complex functional molecules¹⁷ (Scheme 3). For example, treatment of

Scheme 2. Late-Stage Functionalization of Pharmaceutically Relevant Compounds



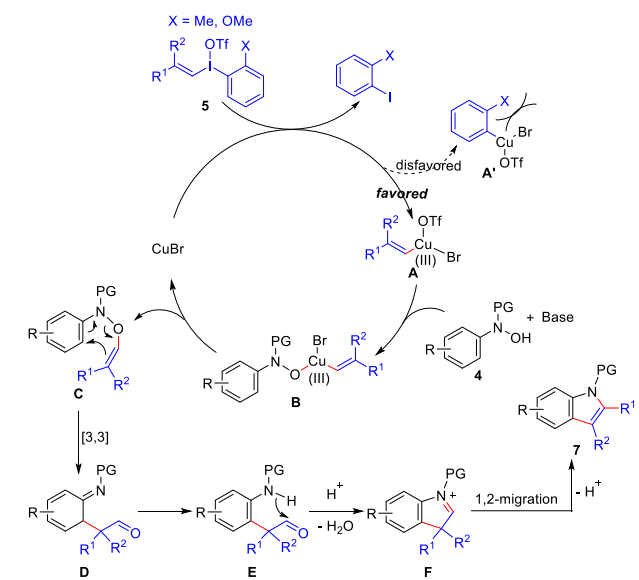
Scheme 3. Synthetic Applications of the Indole Products^{4a}

^aLegend: (a) **9**, Cu(OTf)₂, 1,4-dioxane, 70 °C, 12 h; (b) **11**, Cu(OTf)₂, CH₃CN, 35 °C, 36 h; (c) nitrostyrene, Zn(OTf)₂, toluene, 80 °C, 12 h; (d) NBS, trifluorotoluene, 100 °C, 1 h; (e) TMSCF₃, PhI(OAc)₂, BQ, K₃PO₄, CH₃CN, 85 °C, 12 h; (f) POCl₃, DMF, toluene, reflux, 42 h; (g) anisole, TfOH, CH₂Cl₂, 25 °C, 1.5 h; (h) 1,2-diphenylethyne, Pd(OAc)₂, TBAB, Cu(OAc)₂, DMF, 100 °C, 12 h. Abbreviations: NBS = *N*-bromosuccinimide, BQ = benzoquinone, DMF = *N,N*-dimethylformamide, TBAB = tetrabutylammonium bromide.

3-phenyl-1*H*-indole (**8a**) or 1-butyl-3*H*-benzo[*e*]indole (**8aaa**) with Cu(OTf)₂ and methyl (*E*)-2-oxo-4-phenylbut-3-enoate (**9**) or (*E*)-2-benzoyl-3-phenylacrylonitrile (**11**) led to 9*H*-pyrrolo[1,2-*a*]indoles **10**^{17a} and **12**^{17b} in moderate to good yields, respectively (Scheme 3, paths *a* and *b*). A Zn(OTf)₂-catalyzed Friedel–Crafts C2-alkylation reaction of 3-substituted indole **8ai** with nitrostyrene as alkylating agent generated C2-alkylated product **13** in 85% yield^{17c} (Scheme 3, path *c*). The C2 position of our indole products can be also readily converted into other important building blocks via various direct C2-functionalization reactions: for instance, bromination,^{17d} trifluoromethylation,^{17e} and formylation^{17f} reactions (Scheme 3, paths *d*–*f*). A TfOH-promoted umpolung hydroarylation reaction of the indole **8aaa**-Ac with anisole was successfully accomplished to afford **17** in a good yield of 70%^{17g} (Scheme 3, path *g*). Additionally, a palladium-catalyzed dehydro-genative annulation of 3-aryl-substituted indole **8ai**-Me with 1,2-diphenylethyne was conducted to produce dibenzo[*c,g*]carbazole **18** in good yield^{17h} (Scheme 3, path *h*).

On the basis of our findings and the previous studies related to the combination of copper catalysts and iodonium salts which have been established by Gaunt,^{14a,c,e,g,18} MacMillan,^{14d,f,h} and others,^{13g,19} we proposed a reaction pathway involving a vinyl–Cu^{III} species (Scheme 4). We postulated that CuBr will undergo chemoselective oxidative addition into the vinyl–iodine bond to form the highly electrophilic alkenyl–Cu^{III} complex **A** rather than aryl–Cu^{III} complex **A'** because of the *ortho* effect^{13n,20} in the presence of vinyliodonium triflate **5**. The complexation/nucleophilic attack of arylhydroxylamine

Scheme 4. Proposed Mechanism of the One-Pot Process for Indole Synthesis



4 to alkenyl–Cu^{III} complex **A** is expected to generate intermediate **B**, which upon reductive elimination will afford the *O*-vinyl-*N*-arylhydroxylamine **C** and reconstitute the active CuBr catalyst to complete the catalytic cycle. *O*-vinyl-*N*-arylhydroxylamine **C** will undergo a facile [3,3]-sigmatropic rearrangement process, which is similar to the Bartoli indole synthesis, followed by 1,3-proton migration/rearomatization to furnish intermediate **E**. The intramolecular condensation between aldehyde and amide will occur to generate iminium intermediate **F**, and the final indole product **7** will be formed by the 1,2-migration/dehydration/rearomatization of intermediate **F**.

In summary, we have developed a highly efficient copper-catalyzed tandem protocol for the synthesis of substituted indoles and benzoindoles using readily available arylhydroxylamines and vinyliodonium salts under mild conditions. This transformation is tolerant to a broad range of functional groups and provides ready access to a wide selection of indole products in high yield and with excellent regioselectivity. In addition, the indole products can be readily converted into more complex functionalized indoles or polycyclic heterocycles. We envision that this method will be instrumental for the late-stage functionalization of bioactive compounds and drug discovery. Further investigation and synthetic applications are undergoing in our laboratory and will be reported in due course.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acscatal.9b00470.

Experimental procedures, detailed optimization, compound characterization, and NMR spectra (PDF)

Crystallographic data (CIF)

Crystallographic data (CIF)

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Notes

The authors declare no competing financial interest.

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